

**CHEM 203**

**Midterm Exam 1**

October 15, 2013

**ANSWERS**

**Your name:** \_\_\_\_\_

This a closed-notes, closed-book exam

You may use your set of molecular models

**This exam contains 6 pages**

Time: 1h 30 min

1. \_\_\_\_\_ / 15

2. \_\_\_\_\_ / 15

3. \_\_\_\_\_ / 15

4. \_\_\_\_\_ / 15

5. \_\_\_\_\_ / 20

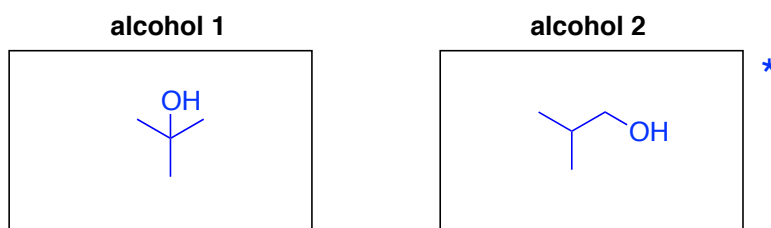
6. \_\_\_\_\_ / 20

**TOTAL** \_\_\_\_\_ / 100

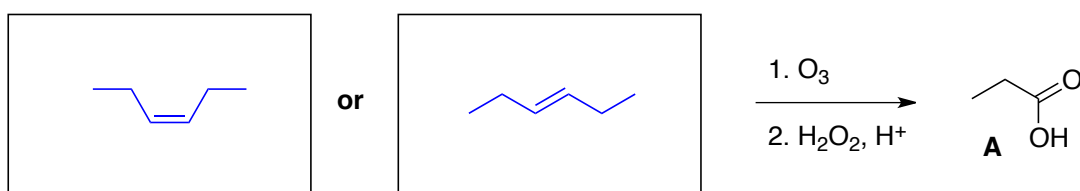
This exam counts for 18.75% of your CHEM 203 final grade

1. (15 pts.) Draw accurate structures of (write your answers in the appropriate boxes):

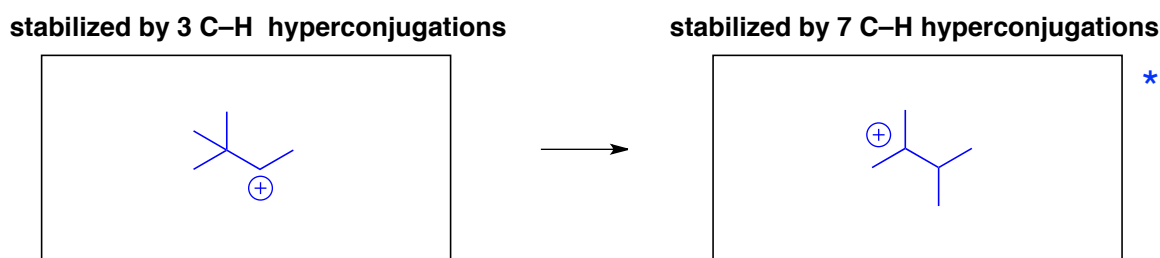
a. Two isomeric alcohols that produce the same alkene upon reaction with  $\text{H}_2\text{SO}_4$  at  $180\text{ }^\circ\text{C}$ :



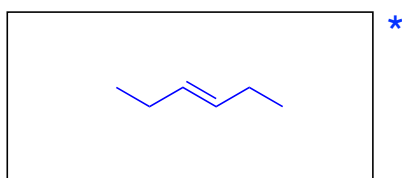
b. Two isomeric alkenes that produce only compound **A** upon reaction with  $\text{O}_3$  followed by  $\text{H}_2\text{O}_2$  and  $\text{H}^+$ :



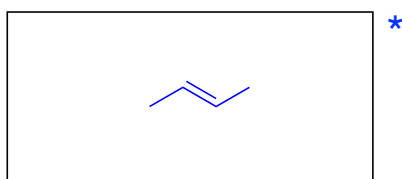
c. A carbocation stabilized by 3 hyperconjugative interactions with C–H bonds that rearranges to give a new carbocation stabilized by 7 hyperconjugative interactions with C–H bonds:



d. An alkene that contains at least 5 carbon atoms and that produces a *meso* dichloride upon reaction with  $\text{Cl}_2$ :

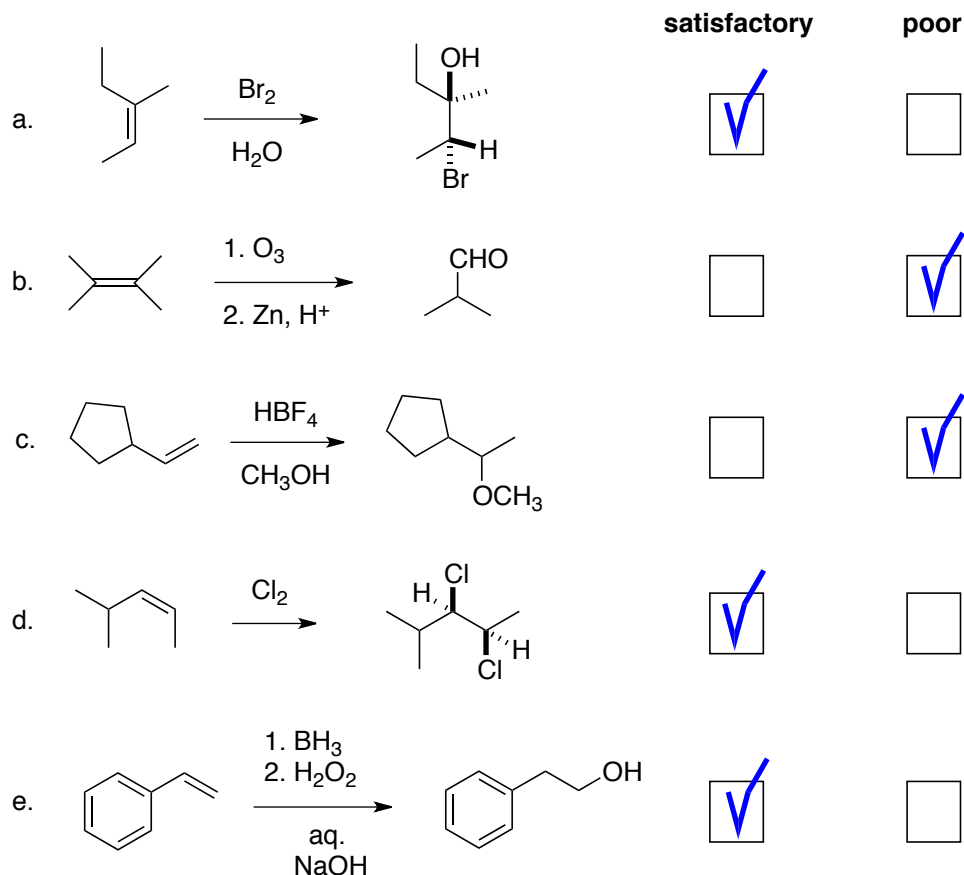


e. An alkene that forms an achiral product upon reaction with  $\text{Br}_2$ , but a chiral product upon reaction with  $\text{Br}_2$  and  $\text{H}_2\text{O}$ :

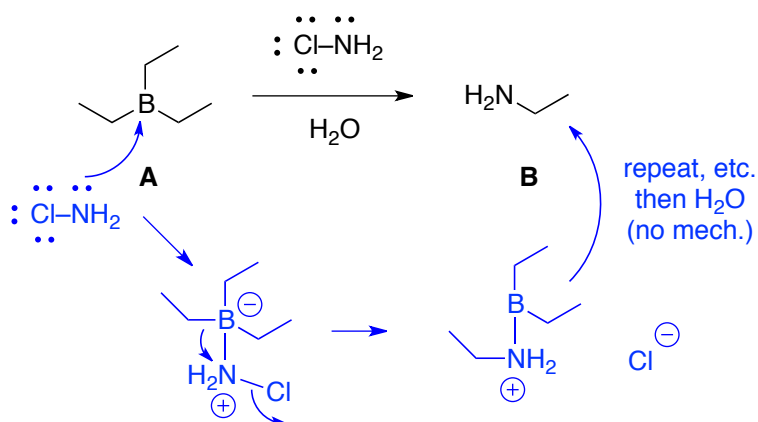


\* other answers are possible

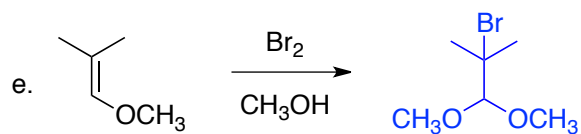
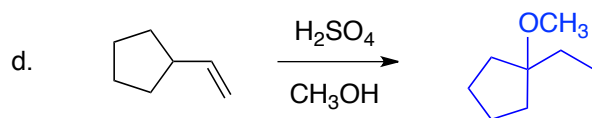
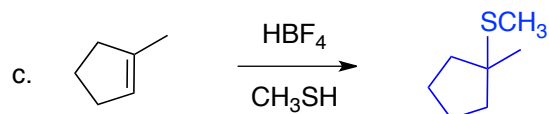
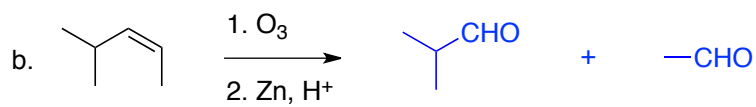
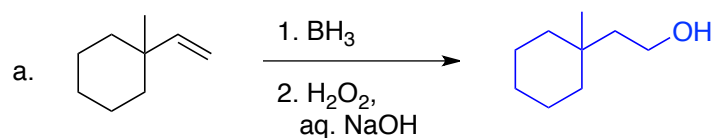
2. (15 pts.) Check the appropriate box to indicate whether the following reactions represent satisfactory or poor methods for the preparation of the products shown:



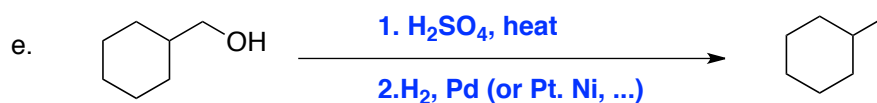
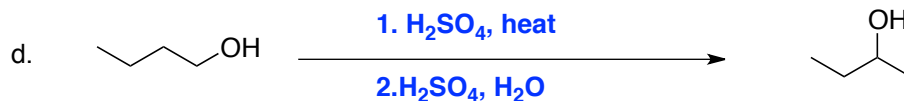
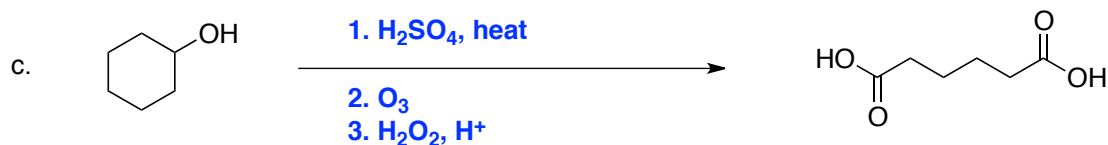
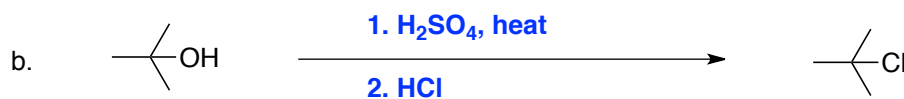
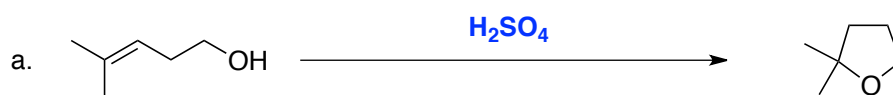
3. (15 pts.) Triethylborane, **A**, reacts with chloramine,  $\text{Cl-NH}_2$ , in the presence of water to give ethylamine, **B**. Write a plausible mechanism (curved arrows, formal charges, etc.) for this reaction (one round only; also, no mech. for the final reaction of the intermediate product with  $\text{H}_2\text{O}$ ):



4. (15 pts.) Provide the structure of the major product(s) expected from the following reactions. If no reaction is expected, answer "NO REACTION". **Important:** compounds incorporating multiple stereogenic centers must be drawn with the correct relative configuration.



5. (20 pts.) Complete the following chemical equations by indicating all reagents / catalysts, in the correct order, that are required to convert the substrates into the products. Provide your answer as a numbered list drawn above / below the reaction arrows. If you should conclude that a product cannot be obtained from the starting compound shown by any method known to you, write "INACCESSIBLE" on the reaction arrow.



6. (20 pts.) Propose a method for the preparation of compounds a. – e. below starting from appropriate alkenes. Draw a clear structure of your proposed starting olefin on the left side of the reaction arrow. Above/below the reaction arrow, list all reagents / catalysts, in the correct order, that are required to induce the desired transformation. **Important:** the desired compound must be the major product of your reaction(s). If a product does not appear to be available by any reaction known to you, write "INACCESSIBLE" on the reaction arrow.

**Note:** it is understood that chiral compounds will be obtained as racemic mixtures.

