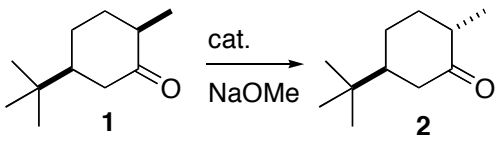
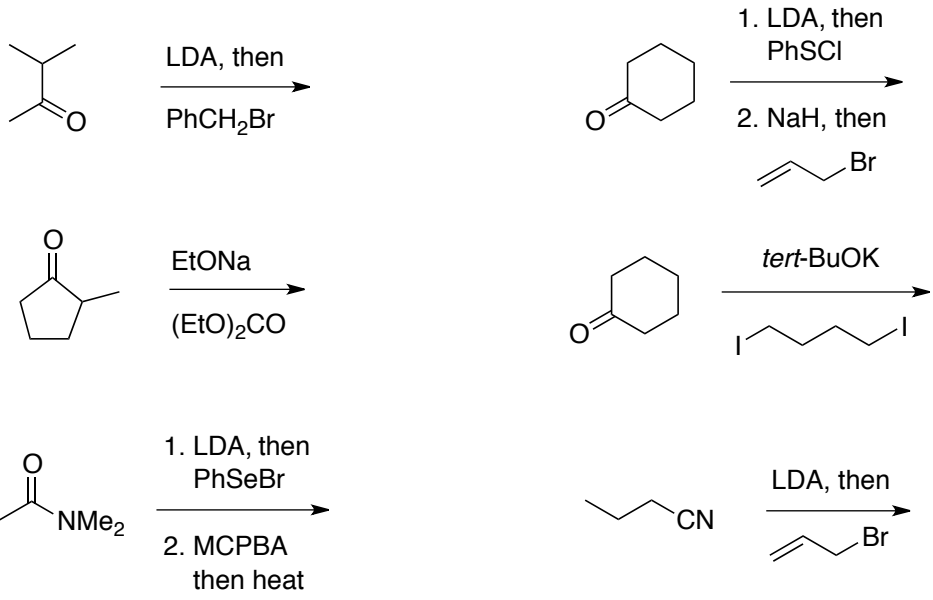
CHEM 330

Problem set 3

1. Provide an explanation for the fact that exposure of **1** to a catalytic amount of NaOMe causes isomerization to **2**.

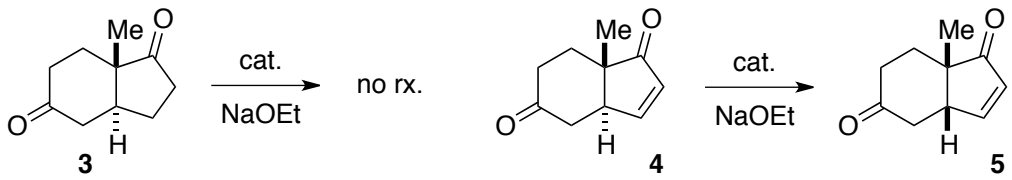


2. Predict the structure of the major product expected from the following reactions:

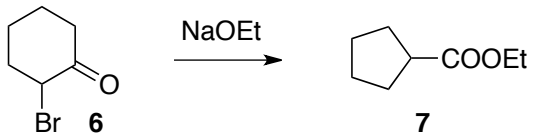


3. Provide an explanation for the following experimental observations and write accurate reaction mechanisms for each transformation:

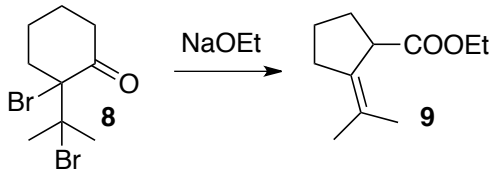
(a) exposure of **3** to a catalytic amount of NaOEt results in no reaction, but treatment of **4** under the same conditions causes isomerization to **5**.



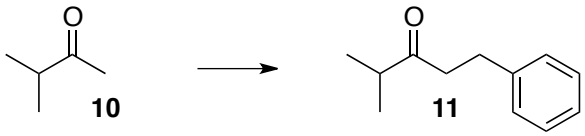
(b) treatment of compound **6** with NaOEt results in formation of **7** (an example of Favorskii reaction)



(b) treatment of compound **8** with NaOEt results in formation of **9**



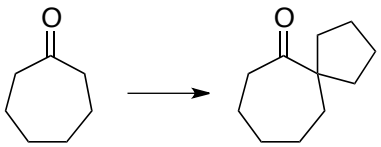
4. In the past, the regioselective alkylation of an unsymmetrical ketone was often achieved through a sequence involving a Claisen-type condensation as a key step. Show how this could be done by proposing an avenue to compound **11** from ketone **10** using any permutation of the solvents/reagents listed below:

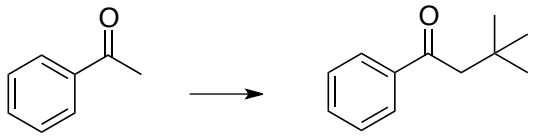


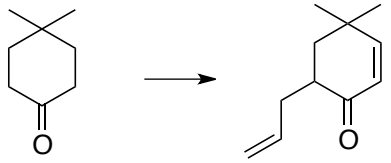
Permissible solvents and reagents:

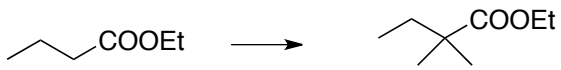
H_2O	NaH	H_2SO_4	CH_3COOEt	MeI
EtOH	KOH	HNO_3	PhCOOEt	$PhCH_2Br$
DMSO	Na_2SO_4	HCl	$(EtO)_2CO$	PhI
Et_2O	$KMnO_4$	TsOH	CH_3COOPh	$Ph(CH_2)_2I$
MeOH	NaBr	H_2CrO_4	PhCOPh	$BrCH_2CH_2Br$

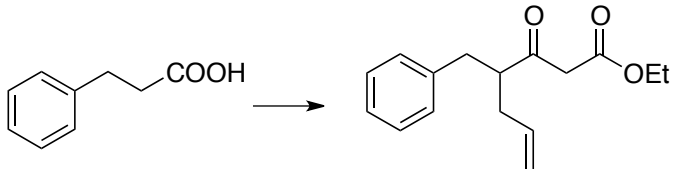
5. Propose a method to achieve the following transformations:

a. 

b. 

c. 

d. 

e. 

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